

# **Raw Materials and Processes in Oleochemistry**

**Markus Dierker**

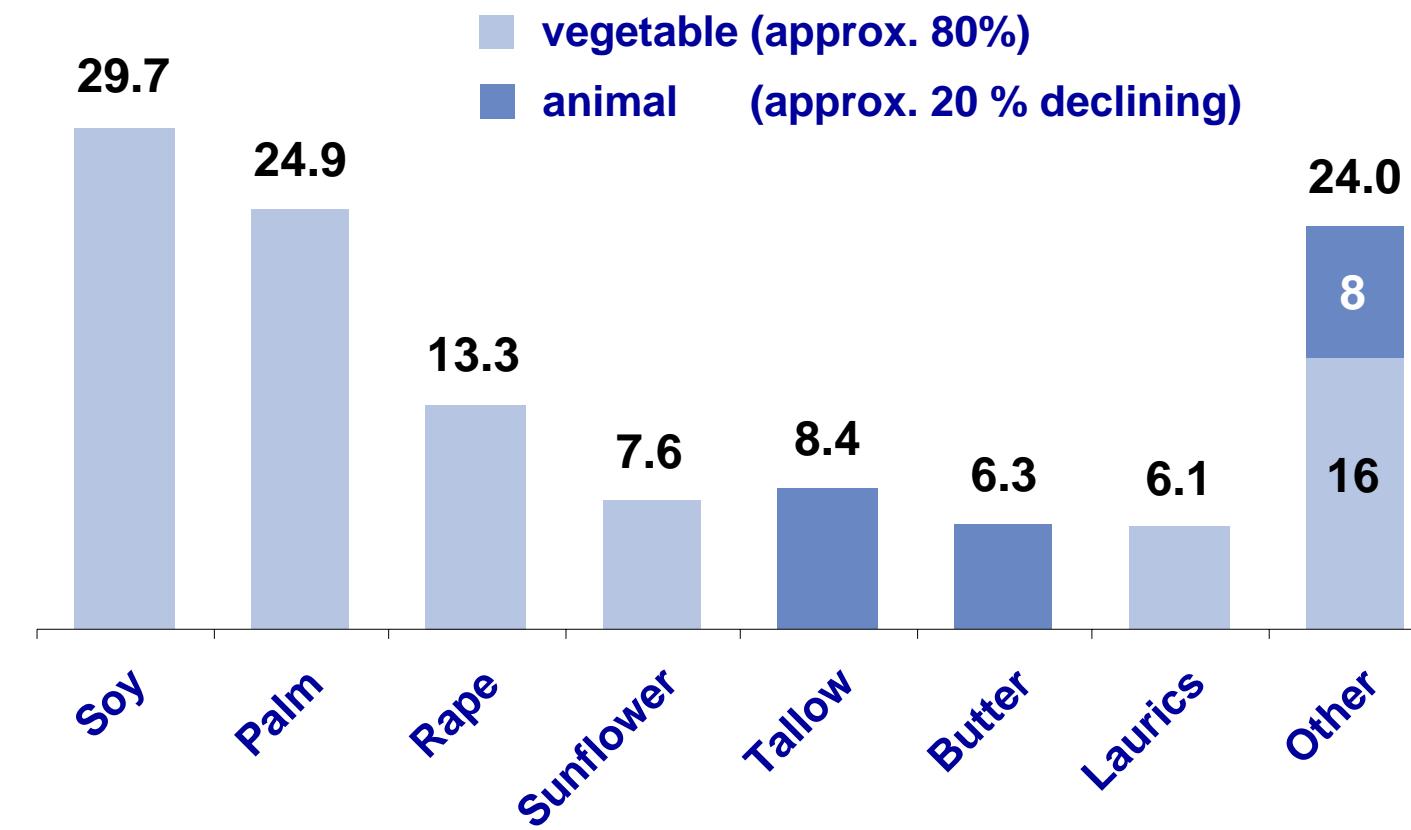
# Contents

- Raw Materials
- Basic Oleochemical Transformations
- Oleochemical Products
- Summary



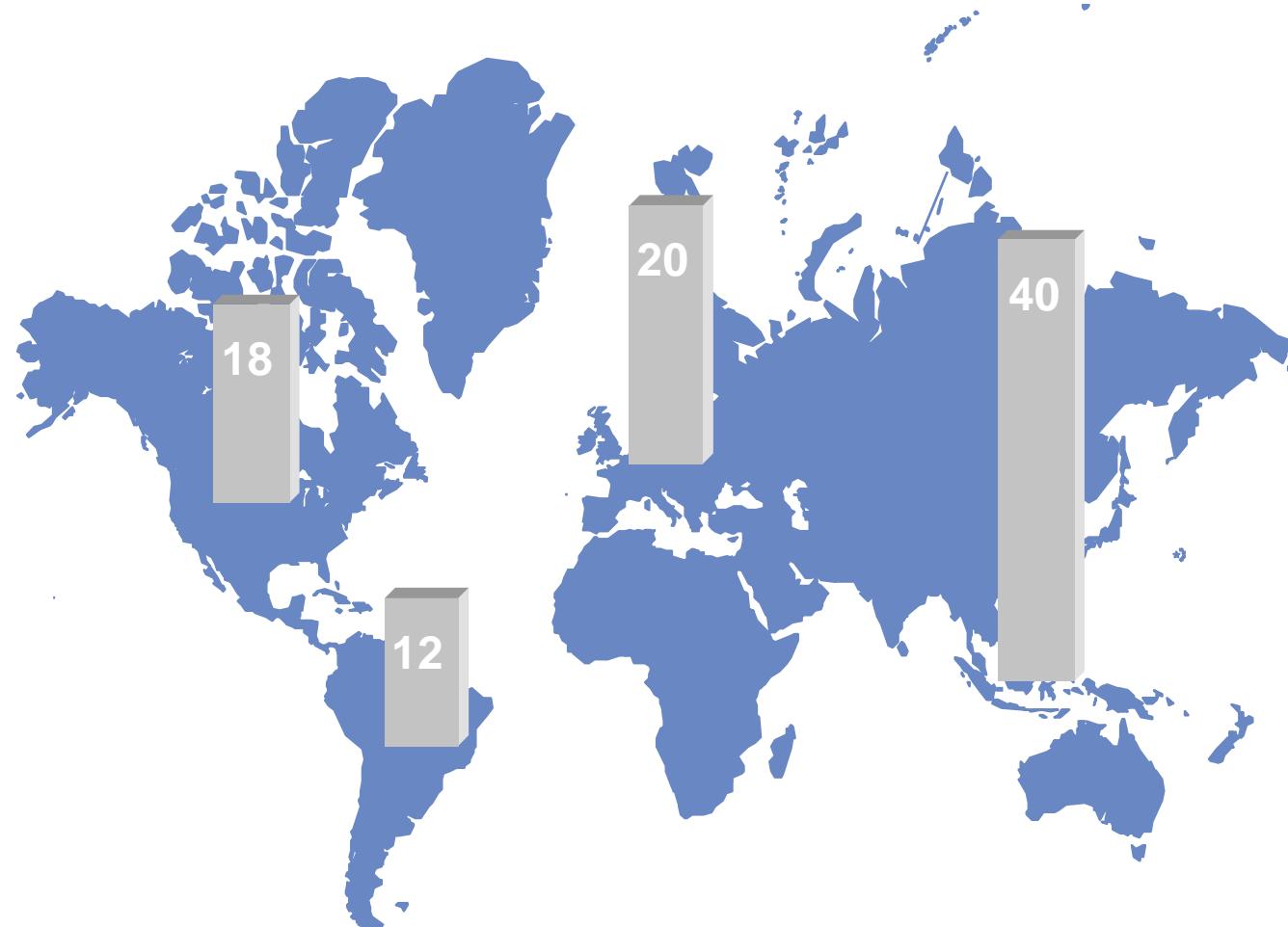
# Raw Materials

## World Production of Oils and Fats 2002 (mil mt)



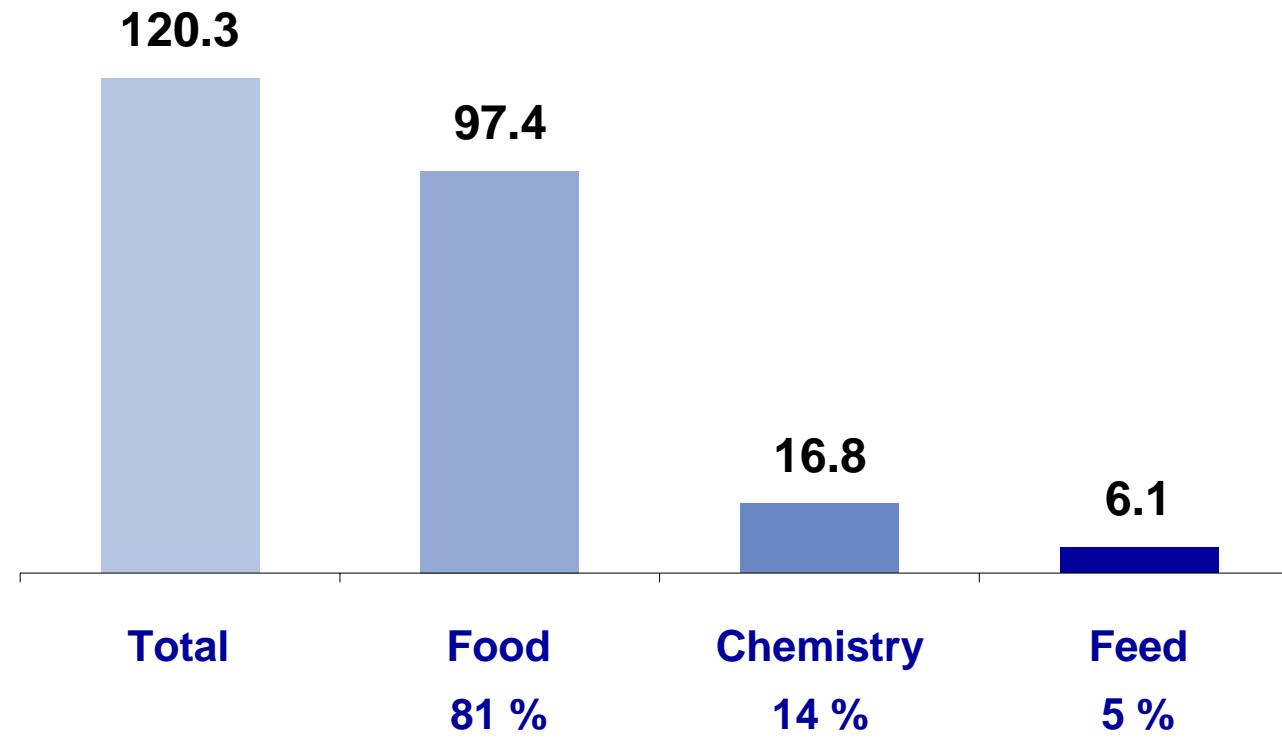
# Raw Materials

## Regions of Production / Share of World Production 2002 (%)



## Raw Materials

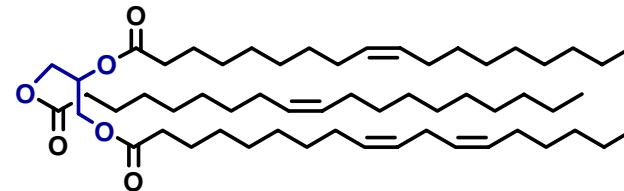
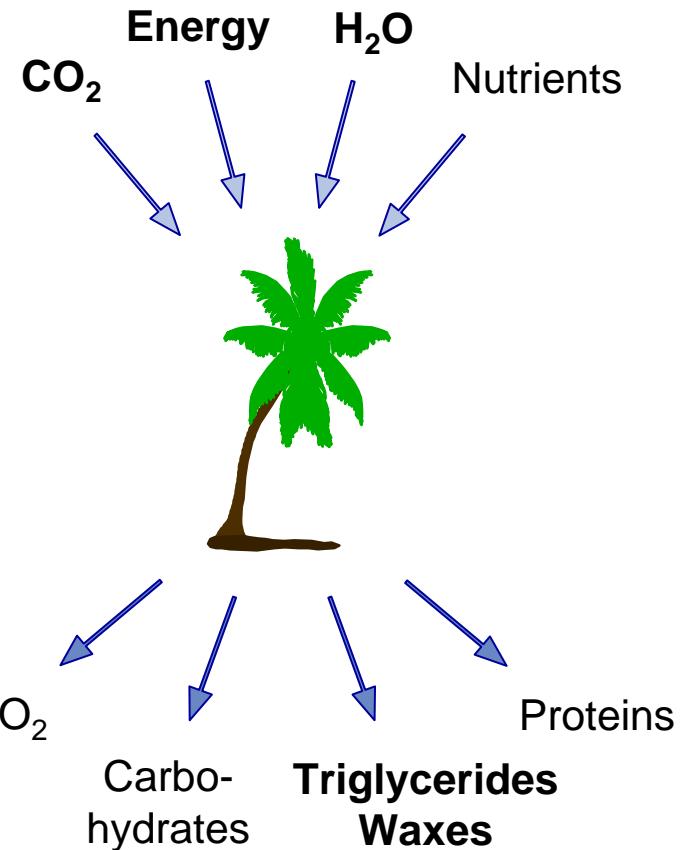
### World Consumption of Oils and Fats (mil mt)



World Consumption of Mineral Oil in 2004: approx. 4.000 mil mt

# Raw Materials

## Natural Generation of Triglycerides

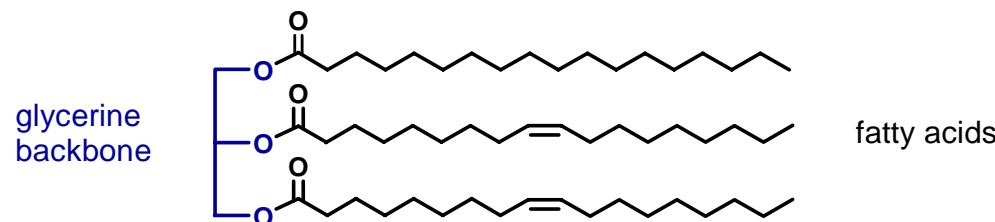


- linear, sat./unsat. FA with even numbered C-chains
- **renewable resources**
- biodegradable
- balanced life cycle analysis



# Raw Materials

## Distribution of Fatty Acids in Triglycerides

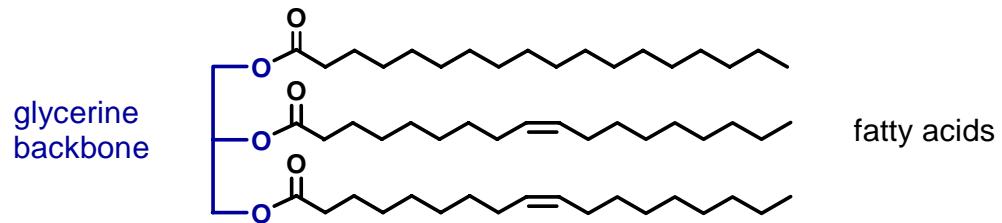


Material	Carbon Chain								C18				
	8	10	12	14	16	18	20	22	0	1	2	3	
coconut oil	8	7	48	17	9	10			2	7	1		
palm kernel oil	4	5	50	15	7	18			2	15	1		
palm oil				2	42	56			5	41	10		
rape oil (old)					2	38	7	51	1	15	15	7	
rape oil (new)						4	90	2	3	1	60	20	9
sunflower (old)						6	93			4	28	61	
sunflower (new)						4	93		1	4	84	5	
soy oil						8	91			4	28	53	6
lard				1	31	65	2		13	46	6		



# Raw Materials

## Distribution of Fatty Acids in Triglycerides



- no statistic distribution of fatty acids in triglycerides

example cocoa butter:    PSO: 36%  
                                 SOS: 25%  
                                 PPO: 15%  
                                 PPS: 1%

P = palmitic acid

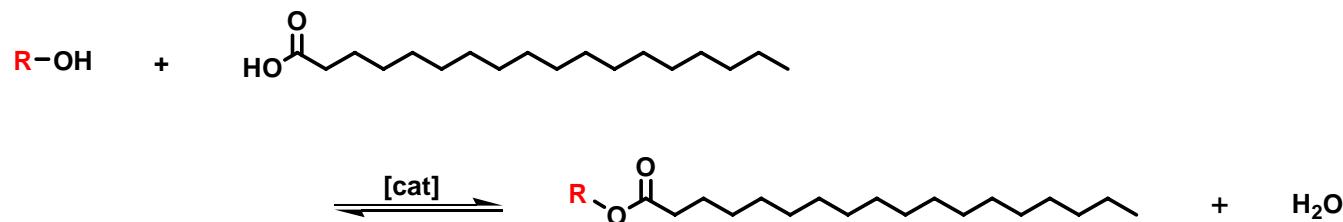
S = stearic acid

O = oleic acid

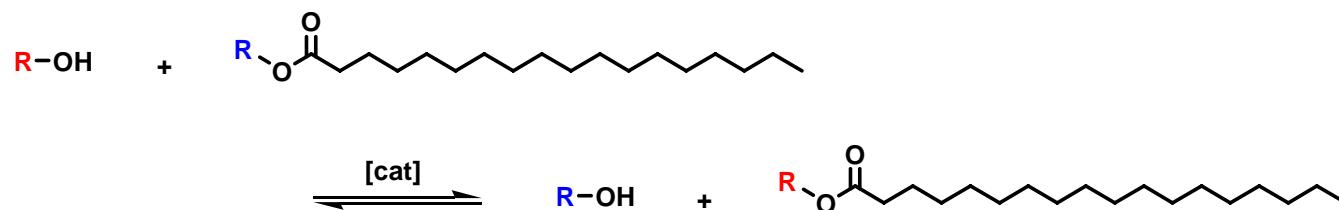


# Basic Oleochemical Transformations

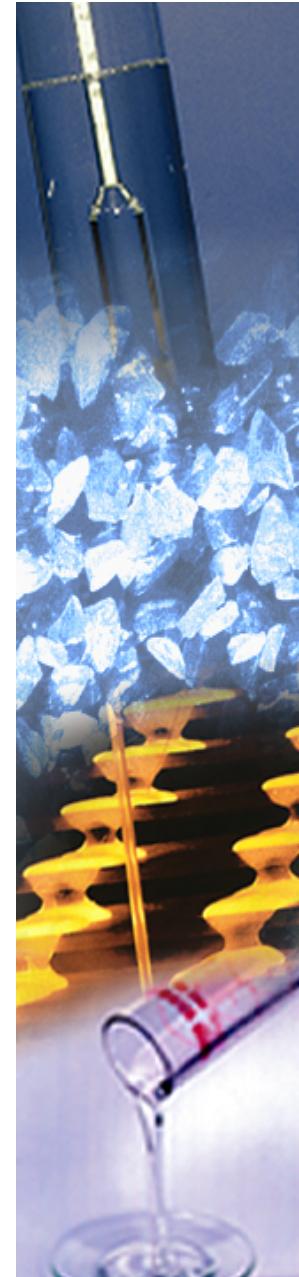
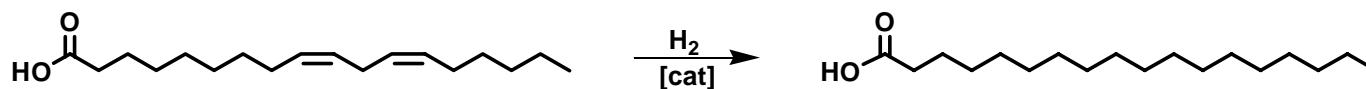
## Esterification / Ester Cleavage



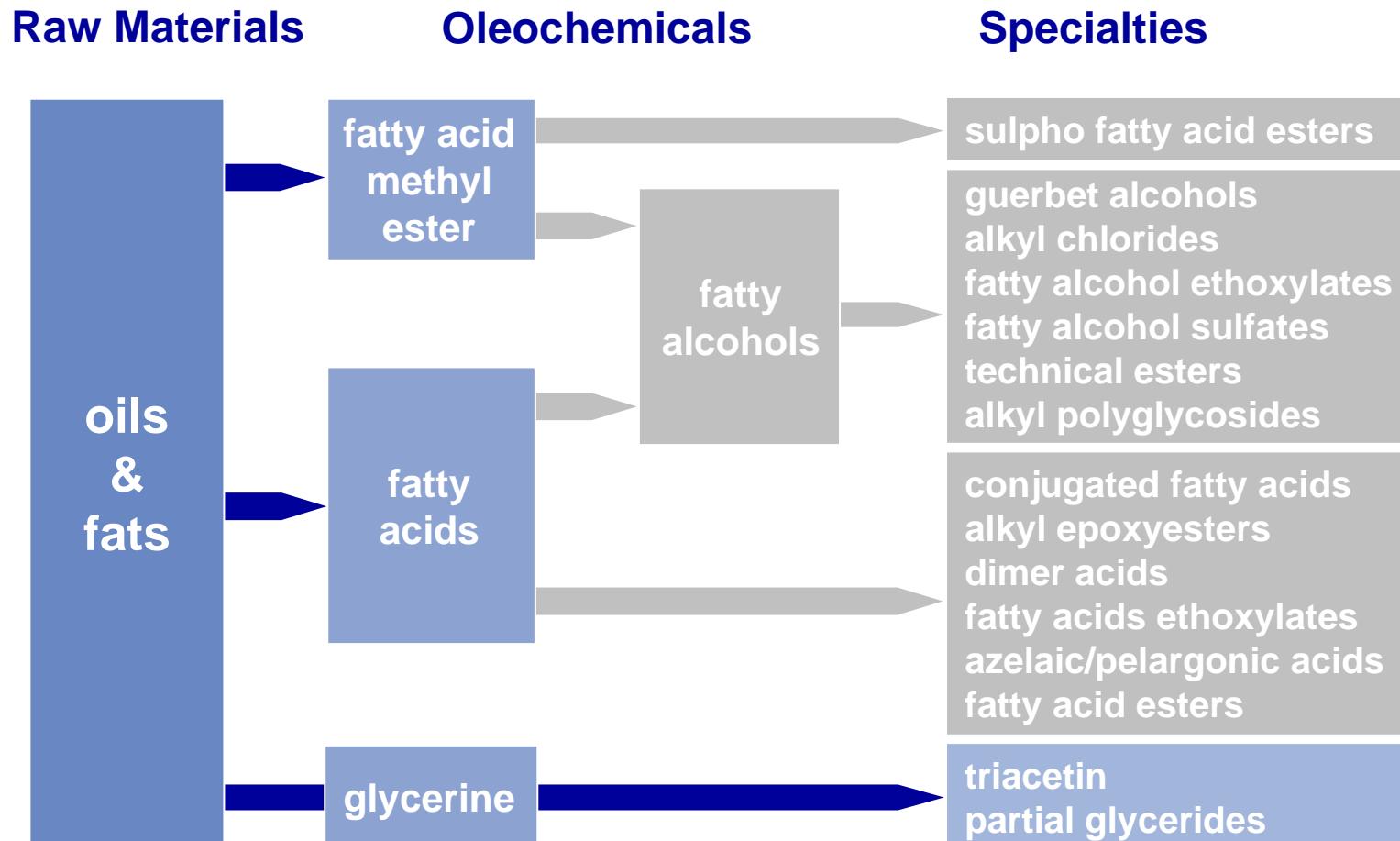
## Transesterification



## Hydrogenation



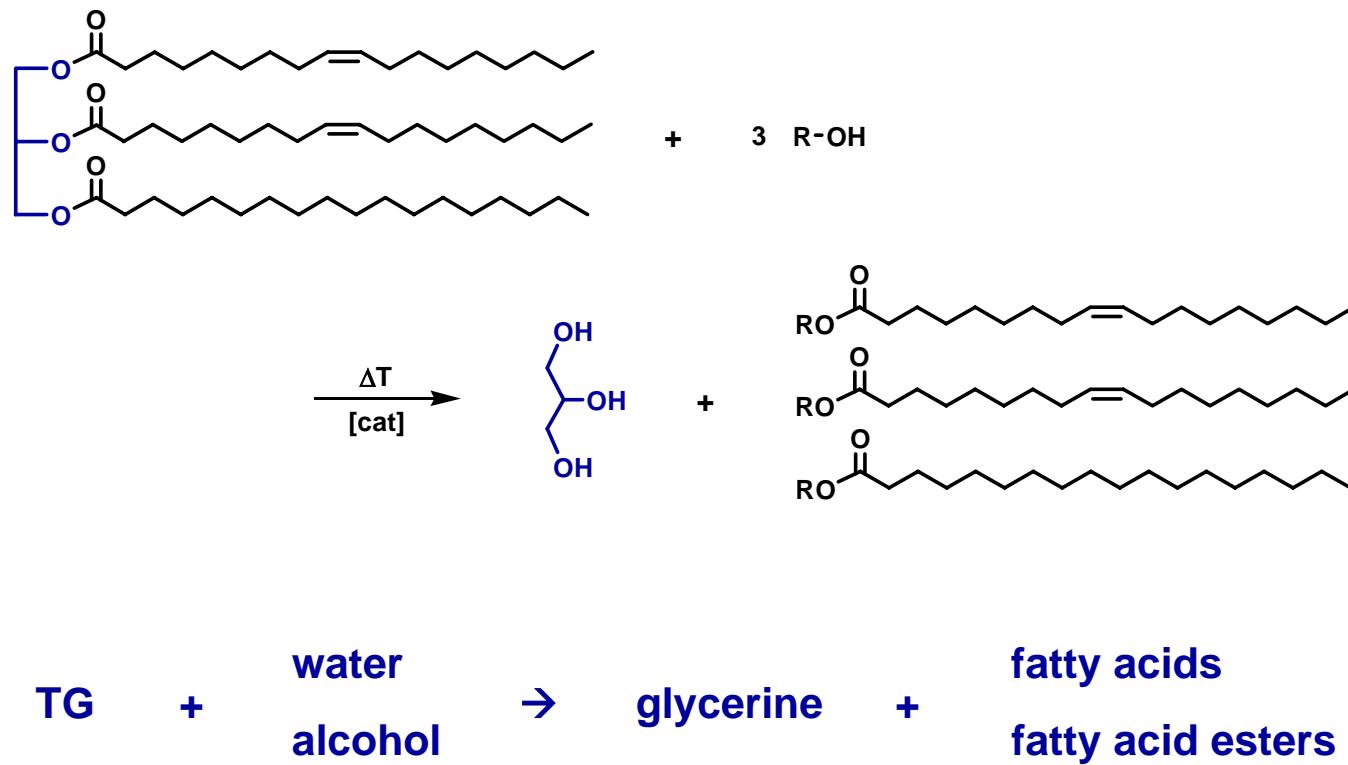
# Production Flow Scheme



# Oleochemicals

## Chemical Conversion of Triglycerides

- **Splitting**

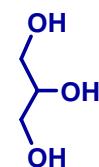


# Oleochemicals

# Glycerine / Derivatives

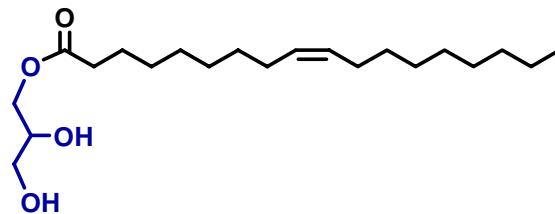
# Glycerine

- purification by distillation or ion exchange
    - pharma-grade glycerine (86%, > 99%)
    - cosmetics, pharma, nutrition, graphic & printing, ...

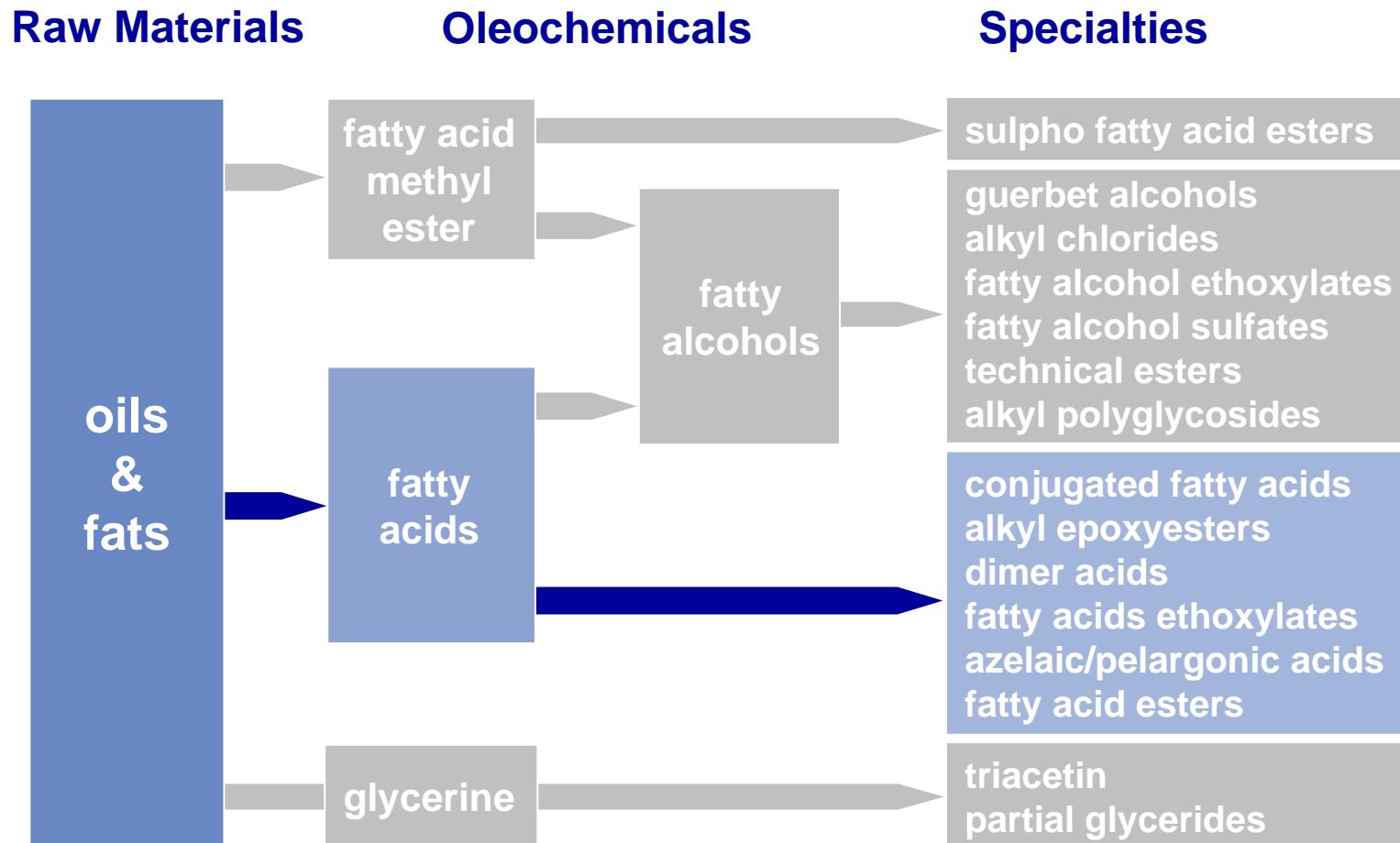


# Glycerides

- esterification / transesterification
    - special mono-, di-, tri-glycerides
      - **GMO** (glycerine monooleate)
      - **triacetin** (triacetyl glycerine)



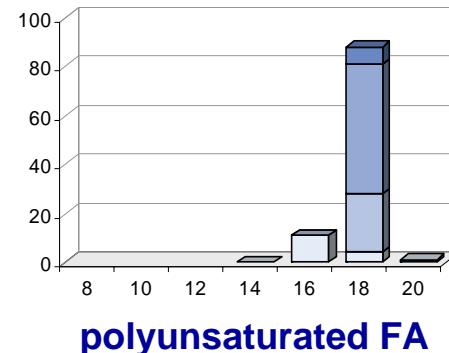
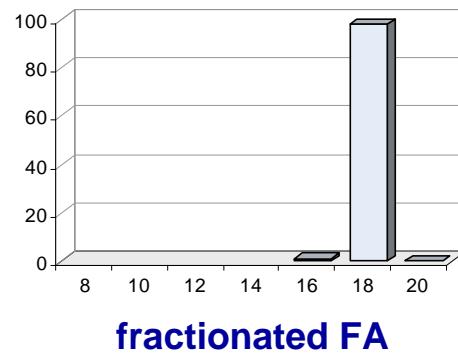
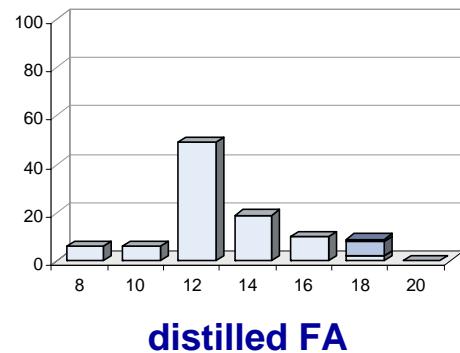
# Production Flow Scheme



# Oleochemicals

## Fatty Acids

Distilled FA	C6-18	broad cuts
Fractionated FA	C8-18	fine cuts, purity > 98%
Stearines	C16/C18	saturated
Oleines	C18	monounsaturated
Polyunsaturated FA	C18	polyunsaturated



→ soaps, detergents, emulsifiers, other oleochemicals, ...



# Oleochemicals

## Fatty Acids - Main Technologies

- Splitting
- Distillation
- Fractionation
- Crystallisation (Hydrophilisation)
- Hydrogenation (partial or full)
- Esterification

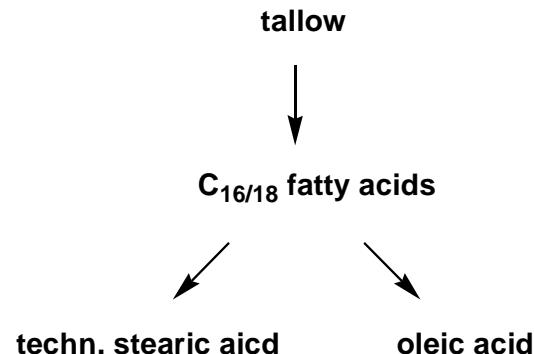


# Oleochemicals

## Fatty Acids – Oleic Acid

- Production

- crystallisation (hydrophilisation)
- saturated acids as by-products



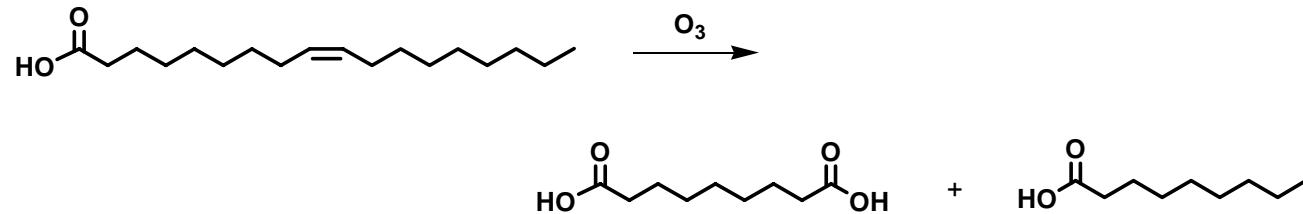
- Properties

- compromise of low melting point and oxidative stability
- melting point: 5 – 8 °C, esters even lower
- relative good stability compared to e.g. rape seed fatty acid due to low amount of polyunsaturated acids
- same behavior in oleic acid derivatives



# Oleochemicals

## Fatty Acids - Specialties



### Azelaic Acid

→ esters, plasticizer, soaps, polyamides, nylon, adhesives, coatings

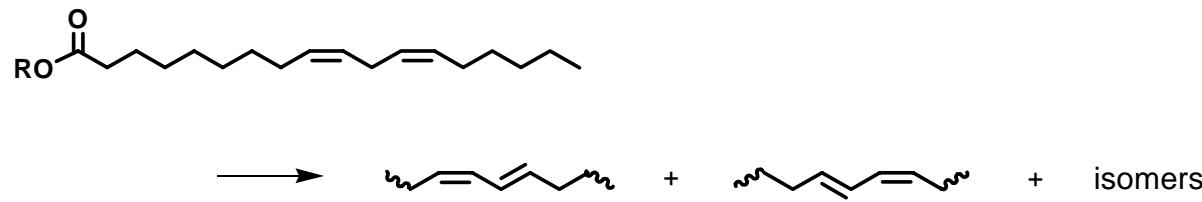
### Pelargonic Acid

→ esters plasticizers, lubricants, corrosion inhibitors



# Oleochemicals

## Conjugated Fatty Acids



### UKD Fatty Acids®

mixture of many isomers  
→ alkyd resins

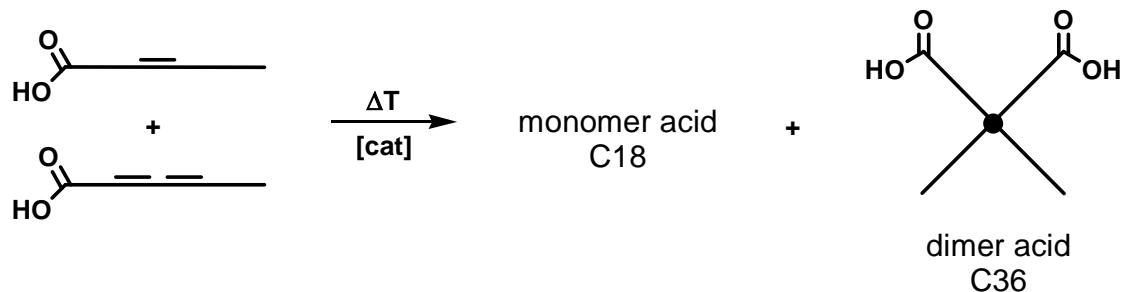
### Tonalin®

two isomers: c9,t11- / t10,c12- CLA  
→ functional foods



# Specialties

## Monomer / Dimer Acids



### Isostearic Acids

### Emersol®

hydrogenation, fractionation → isostearic acid  
→ cosmetics, lubricants, plastic auxilliaries

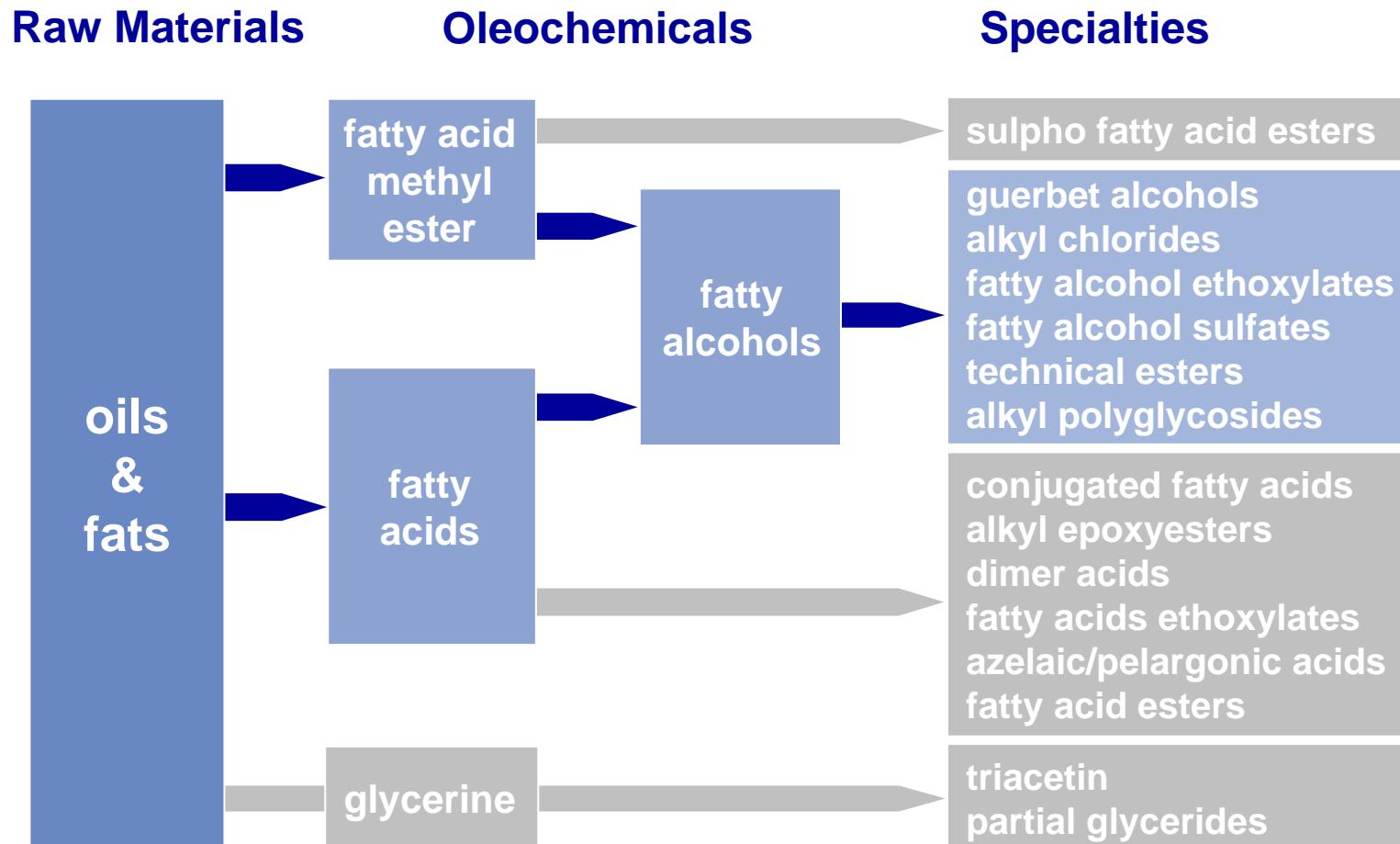
### Dimer Acids

### Empol®

purification  
→ polyamides, polyesters, lubricants, corrosion inhibitors



# Production Flow Scheme



# Oleochemicals

## Fatty Acid Methyl Esters

→ fatty alcohols

→ intermediate to production of special esters

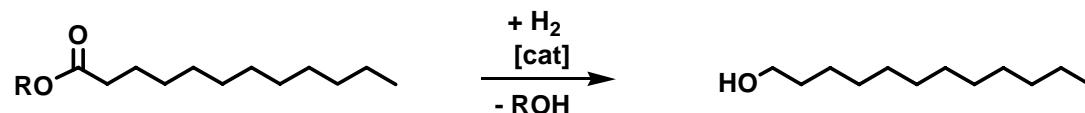
→ solvents, lubricants

→ Biodiesel (rape seed oil methyl ester)



# Oleochemicals

## Fatty Alcohols



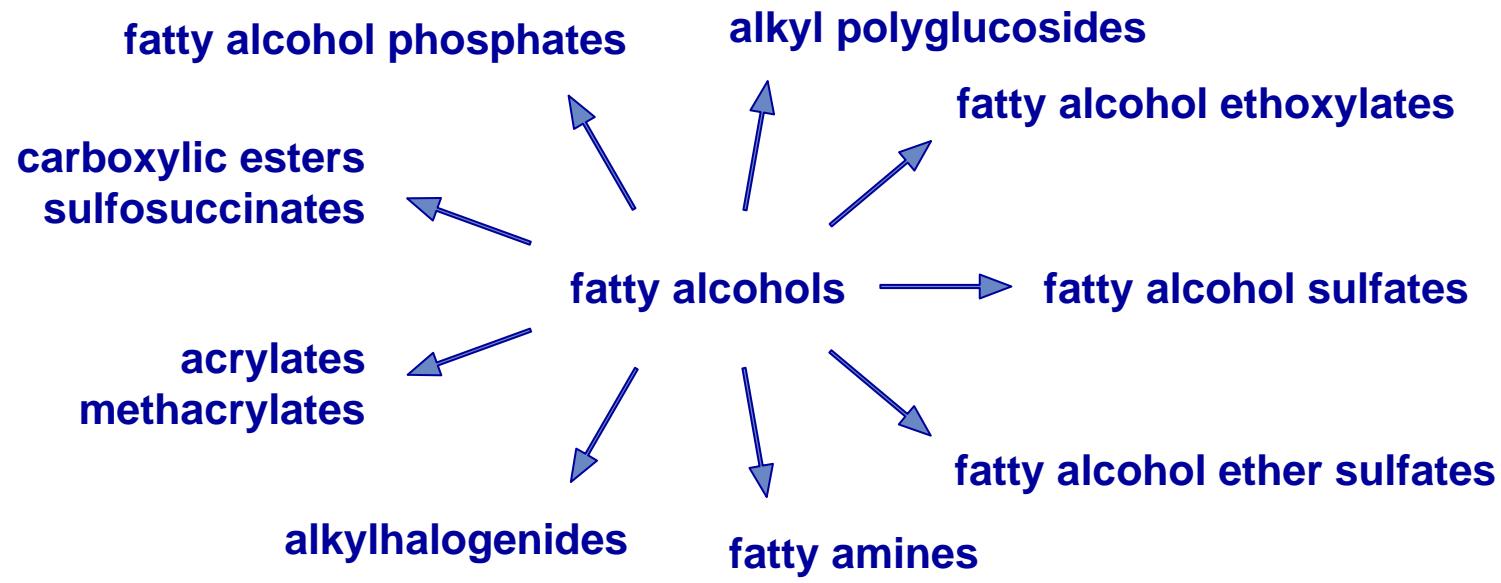
- production by continuous hydrogenation of esters
- over 1 mil mt produced from renewable raw materials
- main raw material for saturated alcohols:  
coconut and palm kernel oil
- competing processes using petrochemical sources
  - Ethylene: Ziegler, Alfol-Process
  - Olefins: Hydroformylation/Reduction
- share of natural sources is rising



# Oleochemicals

## Fatty Alcohols

→ main intermediate to other oleochemicals



→ main application as detergents



# Oleochemicals

## Fatty Alcohols

### Saturated Alcohols

C 6-10 plasticizer range

C12-14 detergent range

C12-18  
C16-22

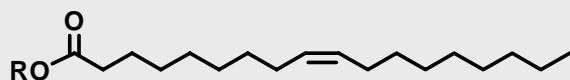
Lorol®, Hydrenol®, Stenol®

solubiliser, defoaming agents, ...

foam stabilisers, lubricant additives, ...

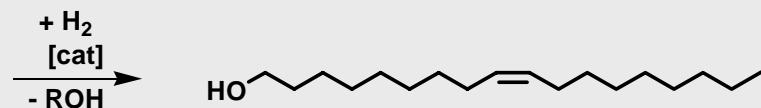
consistency giving factors,  
melting point regulators,  
lubricants for cosmetics and pharma, ...

### Unsaturated Alcohols



C16-18

### HD-Ocenols®



solvents, defoamers, plasticizer,  
emulsifier, oil component in cosmetics



# Oleochemicals

## Fatty Alcohols Specialties

Branched Alcohols

C16-18 satur. / unsatur.

**Speziol®**

detergents  
cosmetics  
fine chemicals

Diols

C8 / C10 / C12 / C18

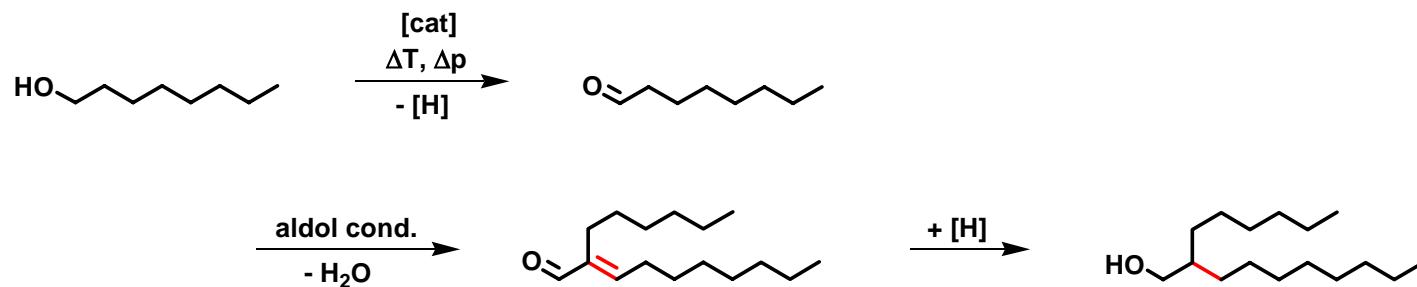
**Speziol®**

cosmetics  
agro chemicals  
pharmaceuticals  
plasticizer



# Specialties

## Guerbet Alcohols



**Guerbet Alcohols**

C16 / C20

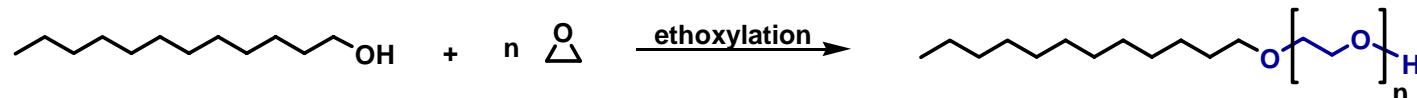
**Rilanit®**

plasticizers  
spray lubrication (metalworking)  
solubilizer for water-miscible MWF  
mold release agents  
oil components in cosmetic application  
dispersing agents



# Surfactants

## Fatty Alcohol Ethoxylates (FAEO)



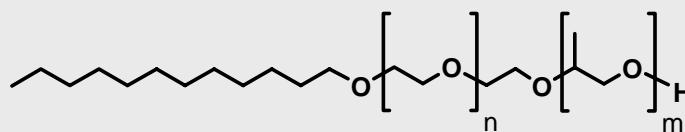
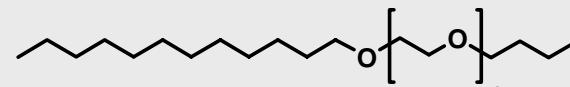
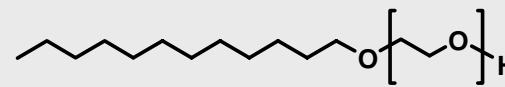
- nonionic surfactant, low foaming
- good emulsifiers for W/O and O/W emulsions
- intermediate to oleochemical products

**FAEO**

**Narrow Range Ethoxylates (NRO)**

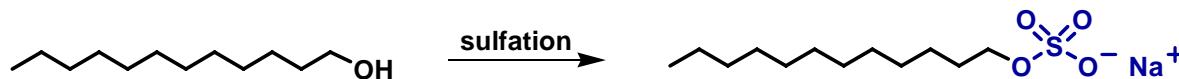
**End-Capped Ethoxylates**

**Fatty Alcohol EO/PO Adducts**



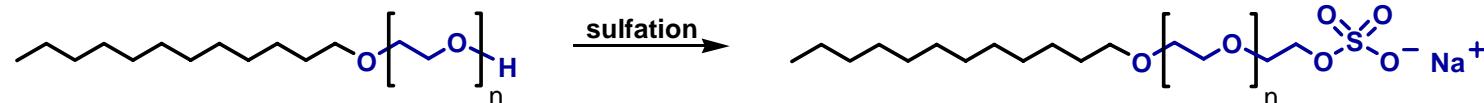
# Surfactants

## Fatty Alcohol Sulfates (FAS)



- anionic surfactants
- good foaming power
- fire extinguisher foams
- detergents for home care and cosmetic application
- insensitive to water hardness

## Fatty Alcohol Ether Sulfates (FAES)

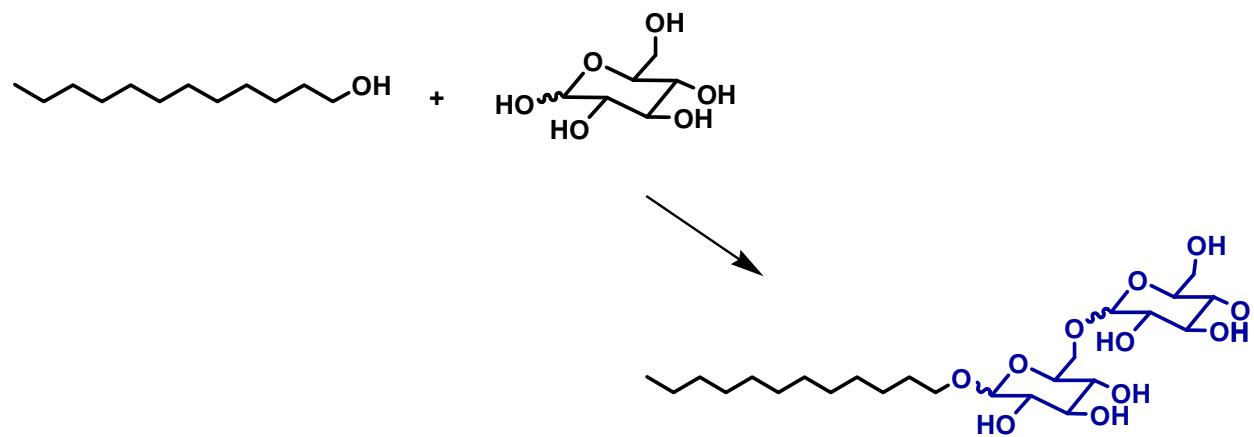


- anionic surfactants
- very good skin compatibility
- very good foaming power
- basic surfactants in cosmetic detergents and cleaners
- insensitive to water hardness
- very good water-solubility



# Surfactants

## Alkyl Polyglucosides (APG)

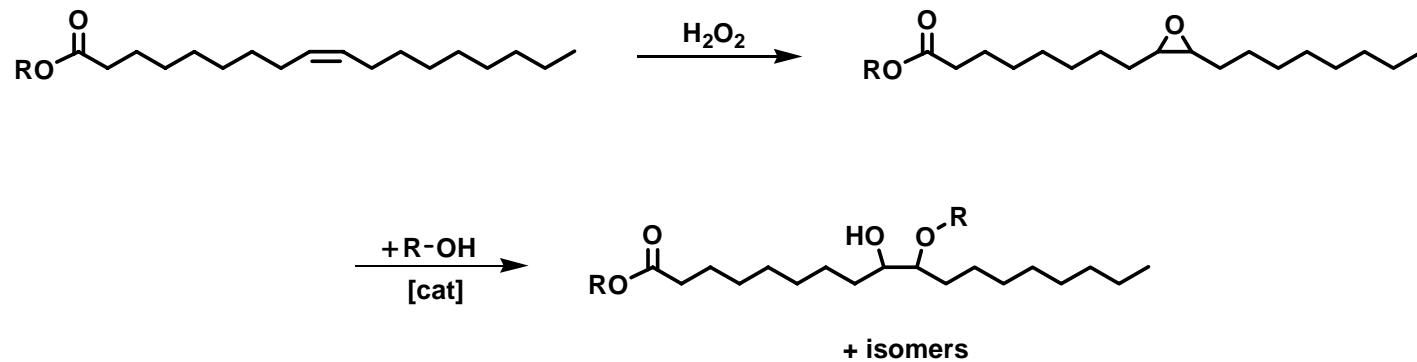


- nonionic surfactant
- very good biodegradability
- detergent for home care applications, cosmetics



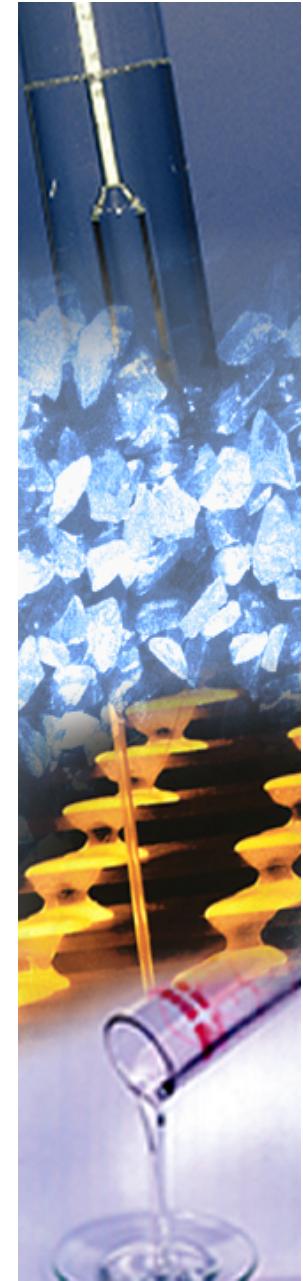
# Specialties

# Polyols - Polymers



- hydrophobic fatty acid polyol
  - adjustable properties / performance

→ self-stabilising softeners, PU components



# Specialties

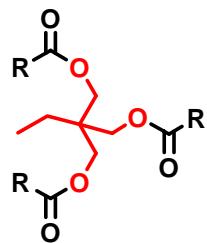
## Polyol Esters - Synlubes

**lubricants, hydraulic fluids, power transmission, MWF, ...**

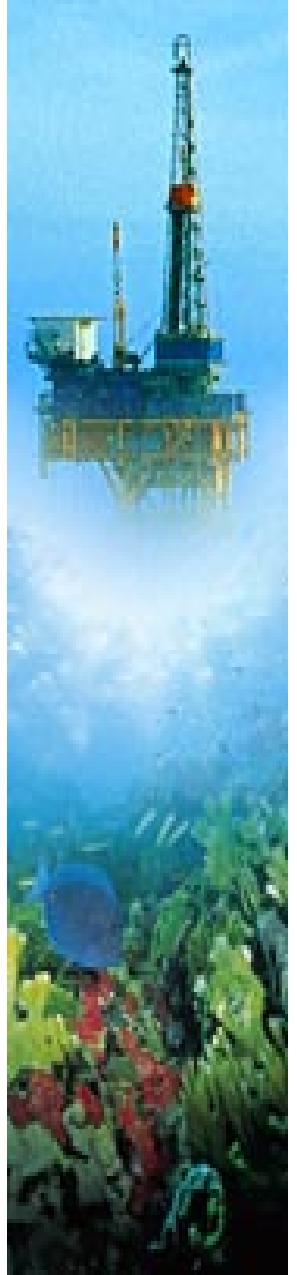
**challenging requirements**

- temperature
- oxidative stress
- long service-life
- price
- viscosity
- CP/PP
- emulsifying properties

## Polyol Esters

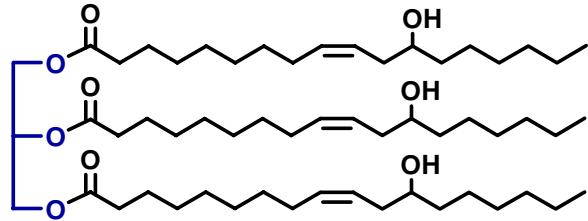


- structural variability
- tunable properties / performance
- biodegradable



# Specialties

## Castor Oil



- mainly produced in India, China and Brazil
- liquid at room temperature
- contains ca. 85% of 12-hydroxy-9-octadecenoic acid

→ triglyceride used as component of polyurethans (often with resins)

→ ricinoleic acid / ricinoleic acid methyl ester

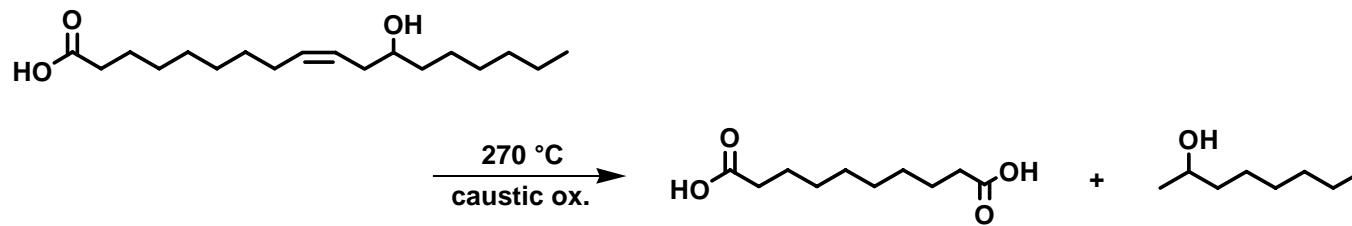
→ 12-hydroxy-stearic acid



# Specialties

## Ricinoleic Acid

→ intermediate to sebacic acid (used for Nylon) and 2-octanol

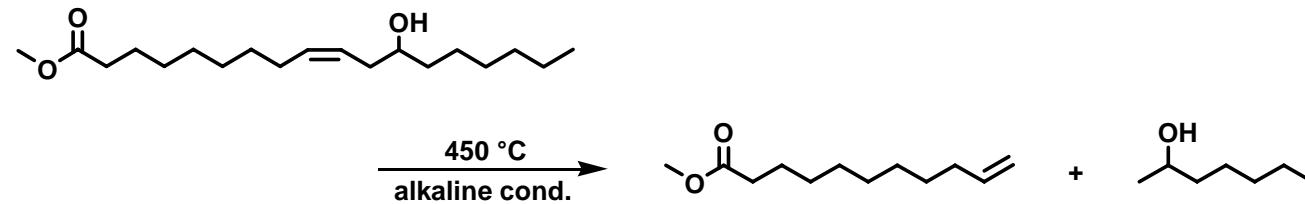


# Specialties

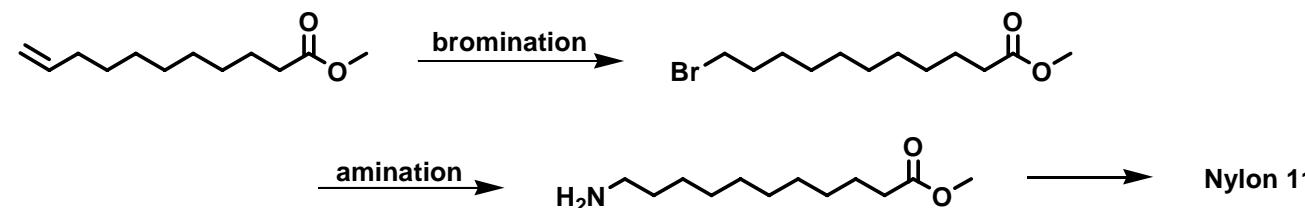
## Ricinoleic Acid Methyl Ester

→ intermediate to undecylenic acid / Nylon 11

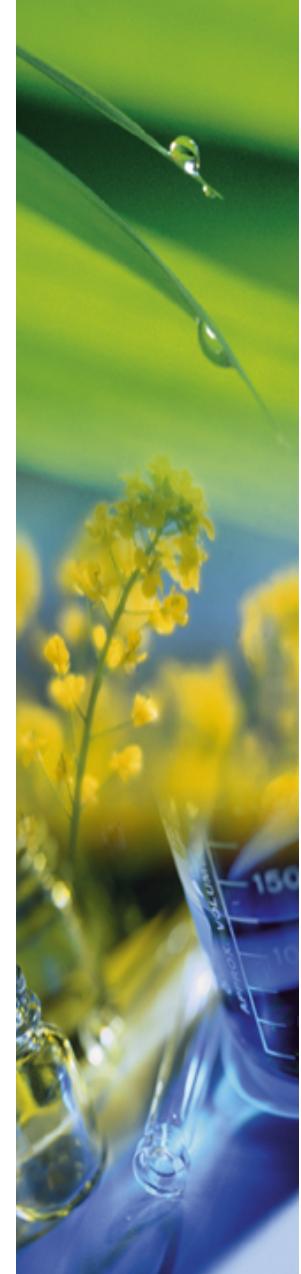
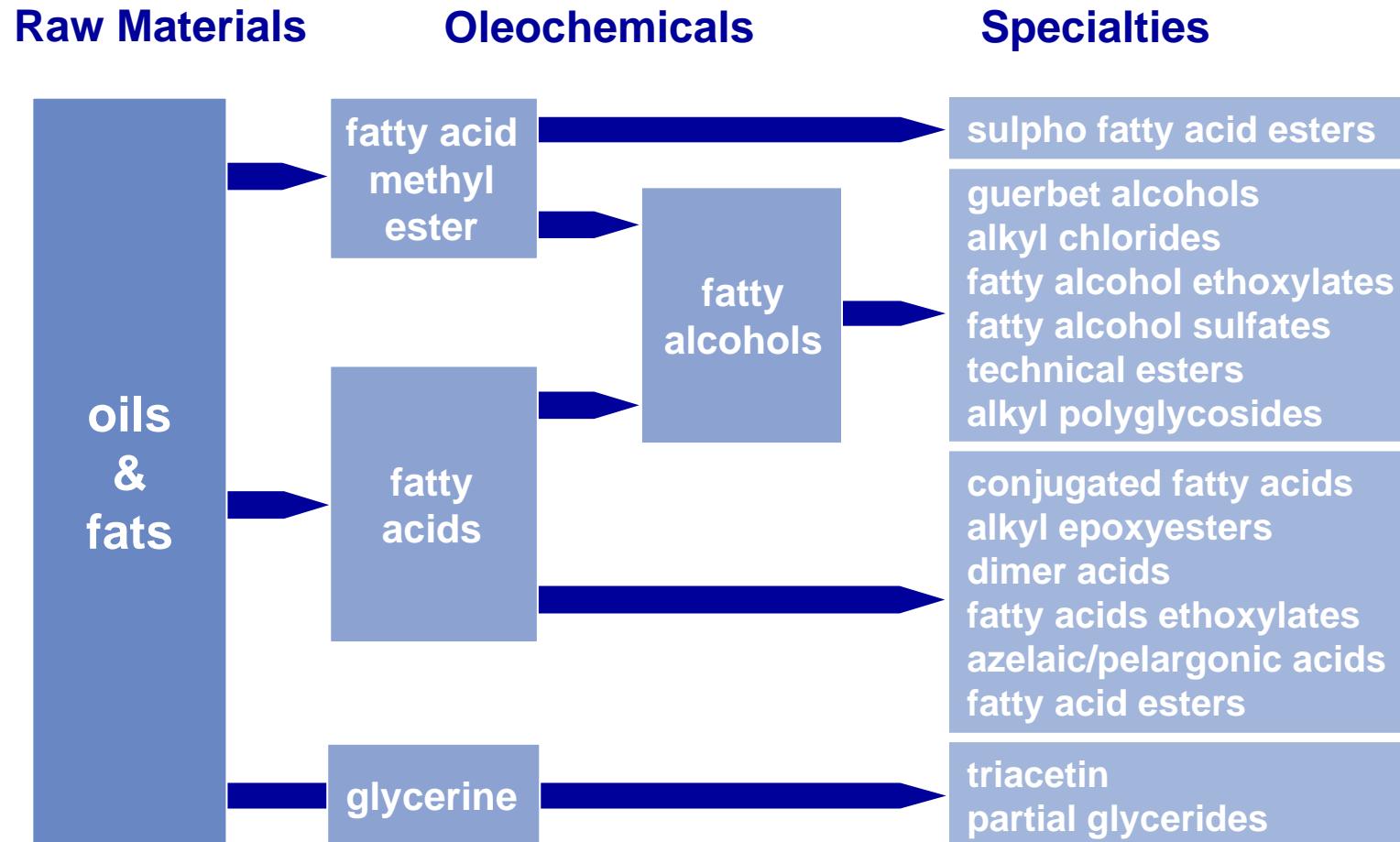
- undecylenic acid methyl ester



- Nylon 11



# Summary - Production Flow Scheme



# Oleochemicals

## Literature

- Encyclopedia of Chemical Technology, 4<sup>th</sup> Edition, 1991,  
Kirk-Othmer
  - Alcohols, higher aliphatic Vol. 1, p. 865
  - Carboxylic acids Vol. 5, p. 147
  - Fats and Oils Vol. 10, p. 252
- Ullmann's Encyclopedia of Industrial Chemistry, 5<sup>th</sup> Edition, 1987,
  - Fatty Acids Vol. A 10, p. 245
  - Fatty Alcohols Vol. A 10, p. 277
- The Lipid Handbook, 1986  
edited by F. D. Gunstone, J. L. Harwood, F. B. Padley  
Chapman and Hall Ltd.



The logo consists of the word "cognis." in white lowercase letters, set against a solid blue rectangular background. The letter "o" is replaced by a detailed blue eye with a black pupil, and the letter "n" has a small white dot at its top right corner.